

253. The Existence of Helical Chirality in a Sterically Congested Naphtho[9]annulenone

Preliminary Communication

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Summary

The existence of the naphtho[9]annulenone **1** in two skeletally enantiomeric forms is demonstrated with the use of ¹H-NMR. spectroscopy in conjunction with a chiral 'shift' reagent.

An interesting feature of cyclic frameworks which are sufficiently large to be nonplanar and to exist in a conformationally mobile state, is that restriction of such mobility should, in the absence of a center of symmetry, lead to the development of skeletal chirality. Work with the unsaturated nine-membered ring [1] has often raised the question as to whether congestive annulation might not sufficiently reduce the system's skeletal flexibility to the point of adopting a rigidly helical shape. The sterically congested naphtho[9]annulenone **1** [2] constitutes a suitable substance, and we examined this possibility by using ¹H-NMR. spectroscopy and a chiral 'shift'-reagent, namely tris (*d, d*-dicamphoylmethanato)europium (III) (Eu(dcm)₃, **2**)²⁾. The reason for employing this composite system is that 'contact'

Table. Chirally induced ¹H-NMR. shifts^{a)} of **1** in the presence of Eu(dcm)₃ (**2**)

Run	Mol ratio [2]/[1]	Signal separation ($\Delta\delta$ (Hz))						
		H _a ^{b)}	H _b	H _c	H _d	H _e	H _f	H _g
1	0.21	16.0	5	4	2	°)	°)	7.7
2	0.25	°)	6	5.8	3.5	°)	°)	9.0
3	0.38	°)	13	11.3	6.0	5.5	°)	18.1
4	0.52	72.9	18	16.4	9	°)	10.6	27.0
5	0.71	96.2	23	21.5	11.9	°)	10.8	35.2

a) 200 MHz in (D₆)benzene.

b) Unequivocal assignment relating to this proton by decoupling procedures in run 5.

c) Shift not available owing to signal overlap.

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²⁾ This substance may be purchased in good purity from the Ventron Division of Alfa Chemicals.

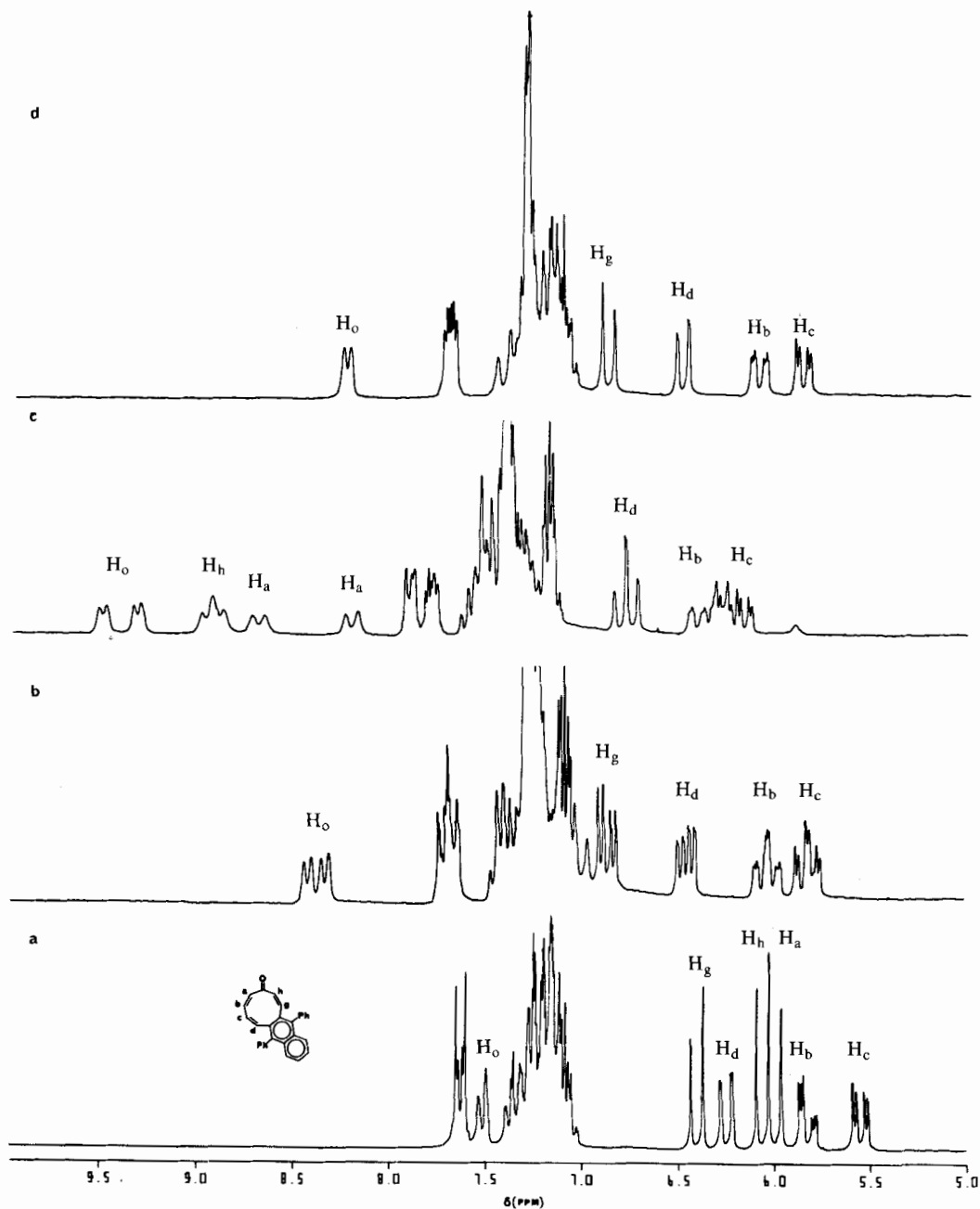
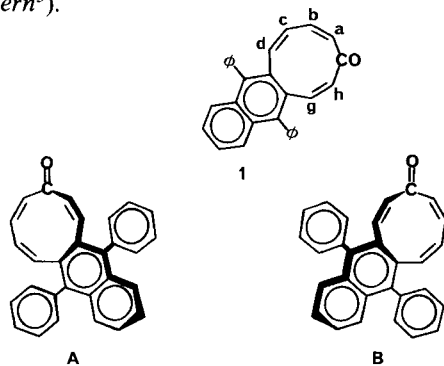


Figure. $^1\text{H-NMR}$ spectra (200 MHz, $(\text{D}_6)\text{benzene}$). a: 1^a ; b: $1+2$ (run 2 in the Table); c: $1+2$ (run 5 in the Table); d: $1+\text{Eu}(\text{dpm})_3$ (molar ratio: $[\text{Eu}(\text{dpm})_3]/[1] \approx 0.27$)

^{a)} Measured coupling constants are as follows: $J(\text{a,b})=12.9$, $J(\text{b,c})=3.6$, $J(\text{c,d})=11.9$, $J(\text{b,d})=1.6$, $J(\text{g,h})=12.6$, $J(\text{o,m})=7.3$.

complexation between the carbonyl O-atom of helical enantiomers **A**, **B** and **2** would lead to two diastereoisomeric arrangements with pairs of chirally related but magnetically distinct protons H_a to H_h .

Our findings (*Table* and *Figure*) establish unequivocally that **1** indeed consists of two chirally related, conformationally rigid forms. The presence of **2** led to substantial separation between *all* enantiomerically related pairs of ring protons yielding the expected patterns (*Figure*, *b* and *c*) and enhancement in signal separation with increased amount of **2**. Particularly revealing in this connection is that the second most pronounced signal separation observed, *i.e.* between the lowest field doublets, measured at 35 Hz in run 5, relates *not* to a proton directly bound to the C_9 -helix but rather to one whose general appearance in the spectrum (*d*, $J = 7.3$ Hz) requires it to be *ortho*-linked (H_o) to one of the phenyl appendages. The fact that such a proton experiences a helicity-induced shift in the $^1\text{H-NMR}$. spectrum means that its phenyl host is subject to restricted rotation. Our interpretation is also strongly supported by the results of a suitable control experiment in which a mixture of **1** and the *nonchiral* shift reagent $\text{Eu}(\text{dpm})_3$ displayed (*Figure*, *d*) the expected downfield $^1\text{H-NMR}$. shifts relative to uncomplexed **1** but with *no change in overall pattern*³).



All attempts at this early stage to estimate key activation parameters for the skeletal inversion process required to interconvert helical enantiomers **A** and **B** were frustrated, not unexpectedly, by the fact that the degree of complexation between **1** and **2** decreases with increasing temperature as revealed by the observation of sharply reduced helicity-induced signal separation as well as overall downfield shifts. When recorded at 80° the $^1\text{H-NMR}$. spectrum of the **1** + **2** mixture listed in (run 3, in the *Table*) displayed features distinctly similar to those of the uncomplexed host (*Figure*, *a*).

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REFERENCES

- [1] *A.G. Anastassiou*, *Acc. Chem. Res.* 5, 281 (1972); *ibid* 9, 453 (1976).
 [2] *A.G. Anastassiou, H.S. Kasmai & M. Sabahi*, *Chem. Commun.* 1979, 1031.

³) Similar results were obtained upon using the *upfield* shift reagent $(\text{D}_{27})\text{Pr}(\text{fod})_3$.